#### **Integrated Research Project**

## Fluoride-salt-cooled High-Temperature Reactor (FHR) with Nuclear Air-Brayton Combined Cycle (NACC)

Integrated FHR Technology Development: Tritium Management, Materials Testing, Salt Chemistry Control, Thermal-Hydraulics and Neutronics with Associated Benchmarking

Charles Forsberg (<u>cforsber@mit.edu</u>)

Molten Salt Workshop 2016
Oak Ridge National Laboratory: October 4, 2016

Massachusetts Institute of Technology University of California at Berkeley University of Wisconsin at Madison University of New Mexico











## **Reactor Basis and Description**

# Much of the Reactor Basis and Technology That is Being Developed is Applicable to all Molten-Salt Reactors

C. Forsberg and P. F. Peterson, "Basis for Fluoride Salt-Cooled High-temperature Reactors with Nuclear Air-Brayton Combined Cycles and Firebrick Resistance Heated Energy Storage", *Nuclear Technology*, 196, Oct. 2016



## **Market Defines Reactor Strategy**

**Understand 2030 Market** 

**Higher Revenue with Variable Electricity Output** 



Power Conversion System to Meet Market Requirements

Base-Load Reactor with Variable Electricity to Grid Using Nuclear Air Brayton Combined Cycle (NACC)



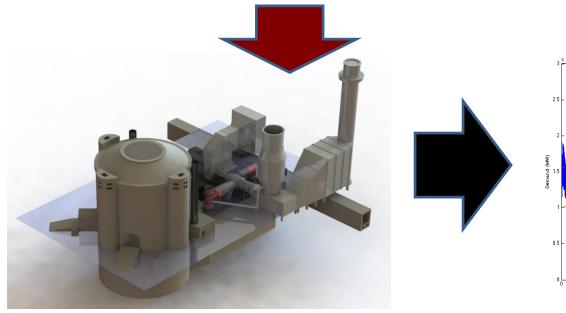
Fluoride-salt-cooled High-Temperature Reactor

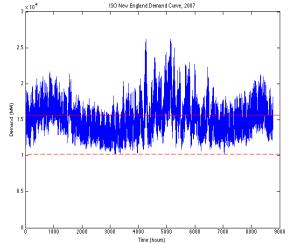


## Fluoride-Salt-Cooled High Temperature Reactor (FHR) with Nuclear Air-Brayton Combined Cycle (NACC)



#### **Stored Heat and/or Natural Gas**





Base-Load Reactor

Gas Turbine

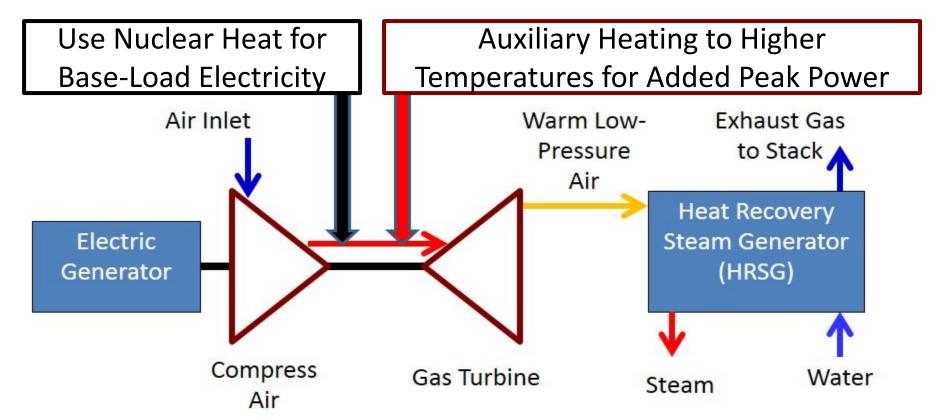
Variable Electricity
And Steam

- 50 to100% Greater Revenue than Base-Load Plant
- Enable Zero-Carbon Energy System when Coupled to Heat Storage
- Safety Strategy to Assure Fuel Integrity in All Accidents

# Nuclear Air-Brayton Combined Cycle (NACC) is a Modified Natural-Gas Combined Cycle

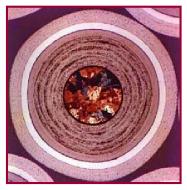
Base-load FHR Heat-to-Electricity Efficiency: 42%

**Incremental Natural Gas-to-Electricity Efficiency: 67%** 

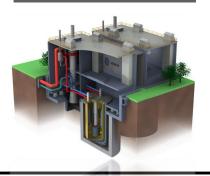


Boost Revenue >50% After Pay for Natural Gas Relative to a Base-Load Nuclear Plant

## **FHR Combines Existing Technologies**







**Fuel:** High-Temperature Coated-Particle Pebble-Bed Fuel Developed for High-Temperature Gas-Cooled Reactors (HTGRs): **Proven Technology** 

Coolant: High-Temperature, Low-Pressure Liquid-Salt Coolant developed for the 1950s Aircraft Nuclear Propulsion Program: Enables Coupling to Gas Turbine; Clean Salt to Minimize Licensing, Corrosion and Maintenance Challenges

Plant Design: Inherited from SFR: Low pressure system, DRACS decay heat removal, Passive shutdown

# IRP Goals Are To Address Major Challenges from Idea to Reactor

- 1. Combining well-known technologies into an innovative concept
- 2. Performing lab-scale experiments to validate computer models
  - 3. Building a new collaboration network to advance FHR technology

4. Developing capabilities to license FHRs and shape future of nuclear power

- Tritium Control and the Role of Carbon (MIT and UW)
- Corrosion Control with Redox Control, Impurity Control, and Materials Selection. (UW and MIT)
- Experiments and Modeling for Thermal Hydraulics, Neutronics and Structural Mechanics (UCB)
- Evaluation Model
   Benchmarking and
   Validation Workshops
   (UCB)



## **Tritium Control (MIT)**

# Lithium Salts Generate Tritium: Must Prevent Tritium Release to Environment

$${}^{6}\text{LiF} + n \rightarrow {}^{4}_{2}\text{He} + {}^{3}_{1}\text{HF}$$

$${}^{7}\text{LiF} + n \rightarrow {}^{4}_{2}\text{He} + {}^{3}_{1}\text{HF} + n'$$

$${}^{19}_{9}\text{F} + n \rightarrow {}^{17}_{8}\text{O} + {}^{3}_{1}\text{H}$$

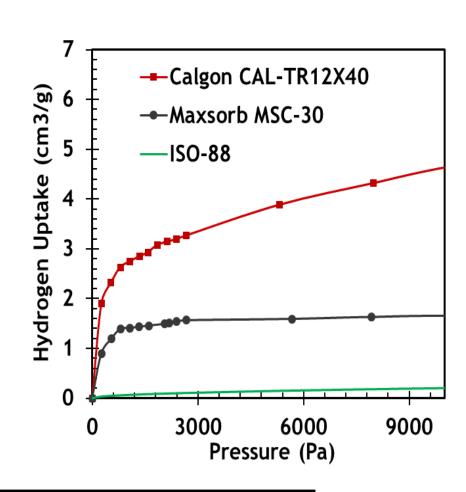
$${}^{9}_{4}\text{BeF}_{2} + n \rightarrow {}^{4}_{2}\text{He} + {}^{6}_{2}\text{He} + 2\text{F}$$

$${}^{6}_{2}\text{He} \rightarrow {}^{6}_{3}\text{Li} + e^{-} + \overline{v}_{e} \quad \left(t_{\frac{1}{2}} = 0.8\text{sec}\right)$$

#### **Tritium Removal from Liquid Salt Using Carbon Beds**

Low-Pressure Measurements Show Large Differences In Hydrogen Sorption For Different Carbons (MIT)

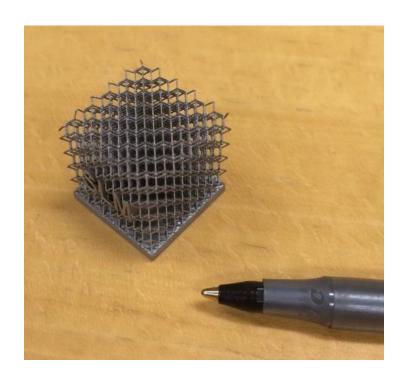
- Carbon (ISO-88) designed for high fluences has low hydrogen sorption at 700° C
- Outside the reactor core one can chose a carbon with high tritium sorption for a tritium removal bed (similar to ion exchange system in an LWR)
- Initial assessment suggests can control tritium levels in FHR with carbon bed external to the reactor core





# Systems Good for Tritium Removal in Clean Salts May Remove MSR Noble Metals

High Surface Area; Good Mass Transfer



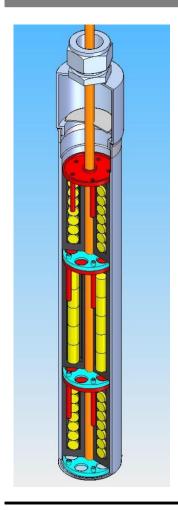
**High-Surface-Area Additive Manufacture Adsorber Bed** 



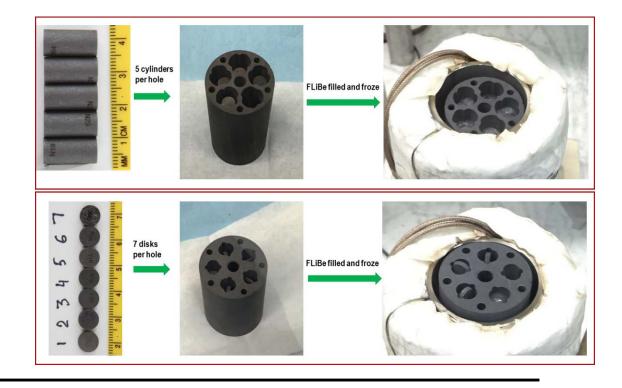
Platinum on High-Surface-Area Carbon (Commercial Catalyst for Hydrogenation Reactions)

## In-Reactor Materials Testing Underway for FHR

3<sup>rd</sup> FHR Irradiation in MITR (Fall 2016)

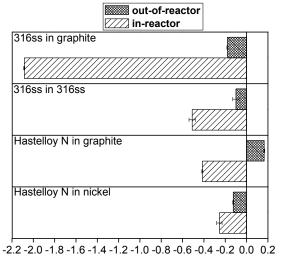


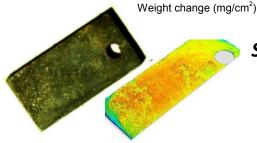
- 1000 hours at 700°C in enriched flibe
- Graphite and C/C specimens (previously irradiated SiC, 316SS, Hastelloy-N, TRISO)



#### Post-Irradiation Exam of FHR Materials

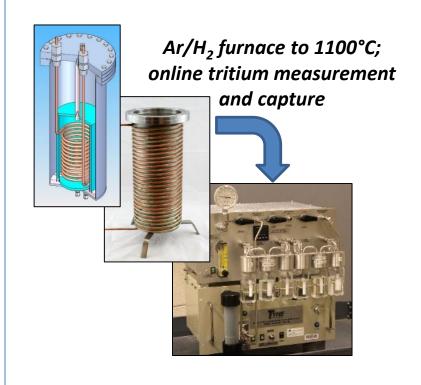
## Irradiation-accelerated corrosion in flibe



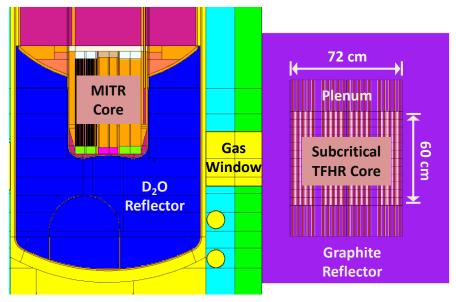


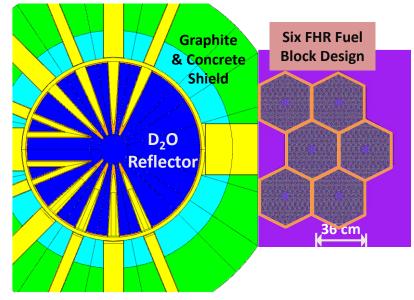
SS316 irradiated in flibe w/ IG-110U graphite crucible

Desorption of tritium from irradiated components: Understanding tritium behavior in an FHR

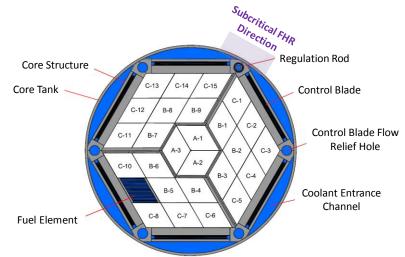


# Developing Reactor-Driven Subcritical FHR Demonstration Option using MITR

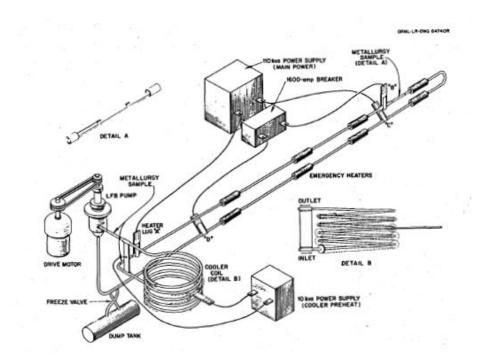


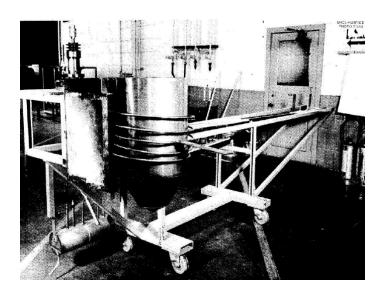


- An MIT Reactor Driven Subcritical System (RDSS) is designed to demonstrate FHR technology.
- ➤ A subcritical system with k<sub>src</sub> of 0.98 is expected to generate 1 MW thermal power ~ 60% average FHR power density.
- Licensed as an MITR experimental facility, not as a new reactor.
- A cost-effective integrated experiment facility suitable for code validation, operation, maintenance, instuments, and component testing. The novel concept reduces risks for licensing a full-scale demonstration reactor.



# Forced circulation flow loops provide the potential next step



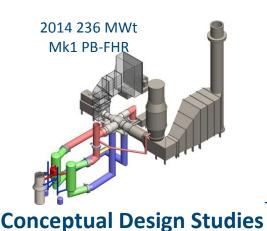


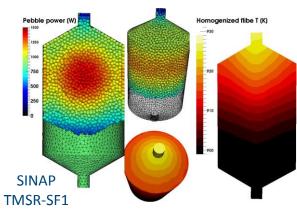
Inserted into reactor beam port

- ORNL forced circulation loop MSR-FCL-1 (ORNL-TM-3866)
- Designed to be inserted into reactor beam port to study irradiation effects.



# UC Berkeley FHR research focuses on thermal hydraulics, neutronics, safety and licensing



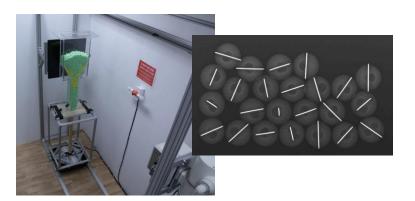


Coupled neutronics and thermal hydraulics



Separate and integral effect tests





X-PREX Pebble Bed Tomography



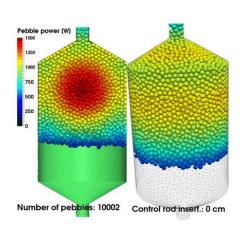


**Organize Expert Workshops and White Papers** 

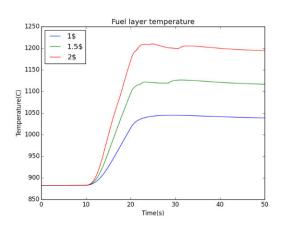
#### Recent UCB FHR Neutronics Advances

- Code-to-code verification is in progress
- Nuclear data uncertainty quantification was performed
- Coupled Monte Carlo/CFD tool was developed for high fidelity (benchmark) calculations
- Parallel development of lower fidelity models for production calculations
- Preliminary results for TMSR-SF1 show that in case of a prompt reactivity insertion, reactivity feedbacks limit the fuel temperature and prevent fuel damage

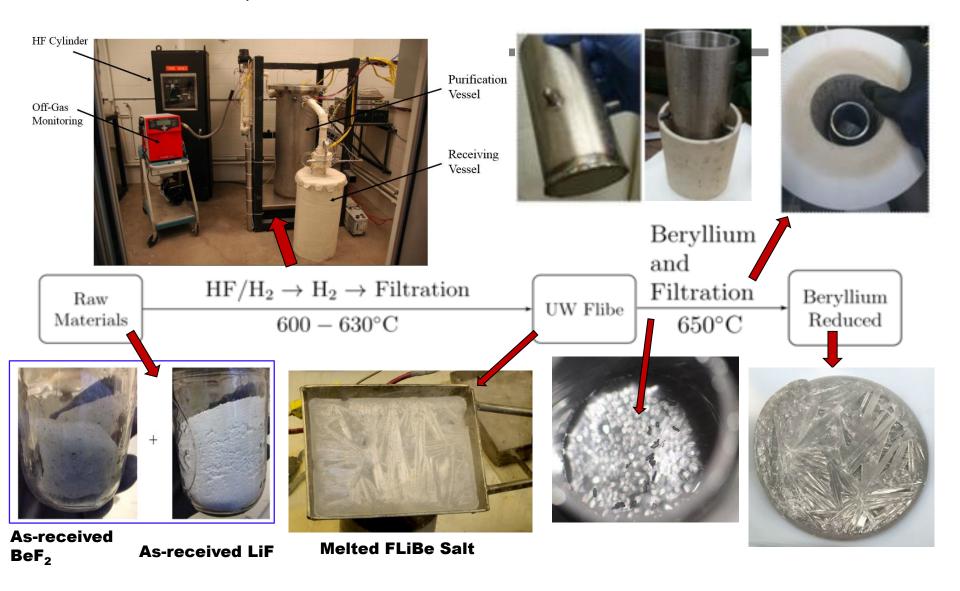
Coupled Serpent-OpenFOAM simulation of control rod insertion in TMSR-SF1



Fuel temperature after a prompt reactivity insertion in TMSR-SF1



## University of Wisconsin - Production, Purification, and Reduction of FLiBe



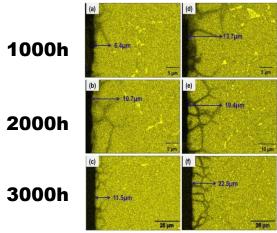
#### **UW Materials Corrosion in FLiBe Salt at 700°C**

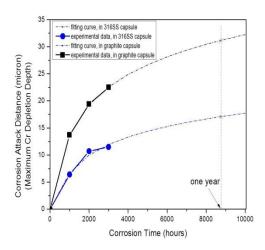
#### Selected Tests Duplicated by MIT with In-Reactor Tests

- Materials Investigated:
  - 316 stainless steel
  - Hastelloy-N
  - SiC-SiC composites
  - C-C composites
  - Graphite
- Additional Materials to be investigated:
  - SiC coated SiC-SiC
  - Diffusion bonded SiC-SiC
  - Mo-Hf-C alloy
  - W-ZrC cermet
- Comparison of corrosion behavior in Be-reduced and unreduced FLiBe

Six Compartment
Graphite Crucible for
Corrosion Tests





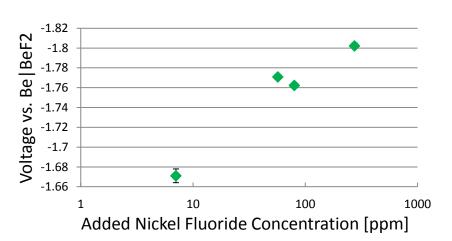


Results for 316 stainless steel tested up to 3000 hours

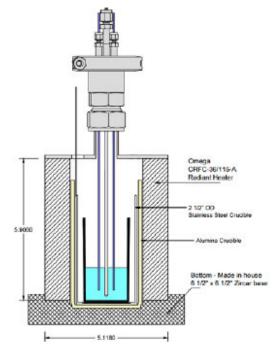
#### **UW Electrochemistry for Redox Potential Measurements**

## Salt Redox Potential: Basis for Understanding and Controlling Corrosion

- A voltage related to the inherent chemical potential energy of the salt
- A measure of a salt's corrosivity
- Useful for understanding results of corrosion experiments
- Determines when chemical reduction of the salt is necessary in order to slow corrosion



Schematic of redox measurement system (right) and Be/BeF<sub>2</sub> reference probe being developed from Afonichkin's (2009) below





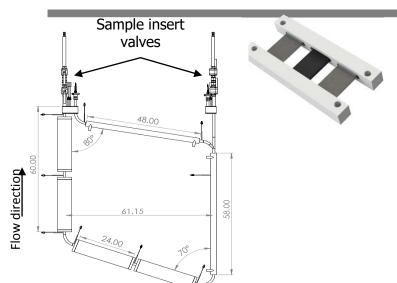


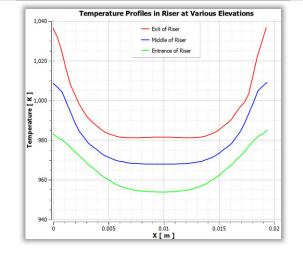
Measurement of the oxidizing effect of metal impurity fluorides on the FLiBe salt redox potential

FliBe electrochemistry globe box

## UW Natural Circulation Molten FLiBe Salt Flow Loop nearly complete

**Enable Measuring Corrosion Under a Wider Set of Conditions** 







Flow-loop schematic and sample holder

**CFD predictions of** temperature profiles at the bottom, middle, and top of the heated riser

IR image during heater testing - inside of the loop is at 700°C

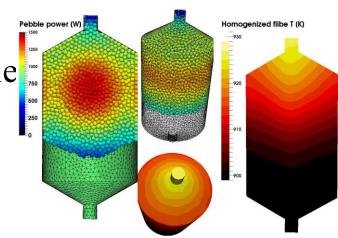
- Thermal hydraulics
  - Flow velocities
  - Temperature profiles
  - Beryllium transport rates
  - Characteristics of the natural circulation
  - Heat transfer characteristics

- Mass Transport
  - Beryllium redox agent transport throughout system
  - Corrosion products transport

- Corrosion
  - Stainless Steel, SiC/SiC, Alloy 800H etc.
  - Flow-assisted corrosion
  - Dissolution in hot leg and plating on cold leg

#### **IRP-SINAP** Interactions

- SINAP 10 MWt test reactor based on IRP design
- Multiple activities at multiple levels
  - Benchmarking
  - Participation in workshops
  - Joint work on tritium control strategie
    - Joint papers
    - Irradiations at MIT of Chinese materials to understand Tritium
  - Exchange of students
- Major university consortium supporting CAS-DOE agreements



UC Berkeley IRP coupled full-core neutronics/TH simulations of TMSR-SF1

#### **Recent Paper Summarizes Basis for FHR**

NUCLEAR TECHNOLOGY · VOLUME 196 · 13-33 · OCTOBER 2016



# Basis for Fluoride Salt—Cooled High-Temperature Reactors with Nuclear Air-Brayton Combined Cycles and Firebrick Resistance-Heated Energy Storage

Charles Forsberg<sup>a\*</sup> and Per F. Peterson<sup>b</sup>

<sup>a</sup>Massachusetts Institute of Technology, 77 Massachusetts Avenue, Cambridge, Massachusetts 02139 <sup>b</sup>University of California, Berkeley, Department of Nuclear Engineering, Berkeley, California 94720

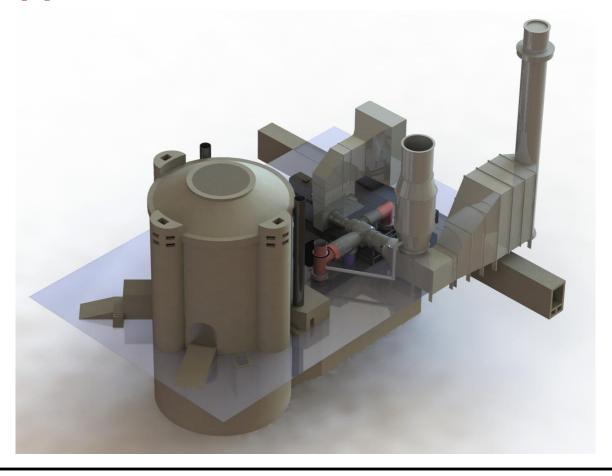
Received January 30, 2016 Accepted for Publication April 30, 2016 http://dx.doi.org/10.13182/NT16-28

Abstract — The fluoride salt-cooled high-temperature reactor (FHR) with a nuclear air-Brayton combined



### **Questions**

IRP Experimental and Analytical Results
Support the FHR and other Salt Concepts





## **Added Information**



## **Biography: Charles Forsberg**

Charles Forsberg is the Director and principle investigator of the High-Temperature Salt-Cooled Reactor Project at the Massachusetts Institute of Technology (MIT). He teaches the nuclear fuel cycle systems and nuclear chemical engineering classes. Before joining MIT, he was a Corporate Fellow at Oak Ridge National Laboratory where he led molten salt reactor studies. He is a Fellow of the American Nuclear Society, a Fellow of the American Association for the Advancement of Science, and recipient of the 2005 Robert E. Wilson Award from the American Institute of Chemical Engineers. He received the American Nuclear Society special award for innovative nuclear reactor design on salt-cooled reactors and the 2014 Seaborg Award. Dr. Forsberg earned his bachelor's degree in chemical engineering from the University of Minnesota and his doctorate in Nuclear Engineering from MIT. He has been awarded 12 patents and has published over 200 papers.



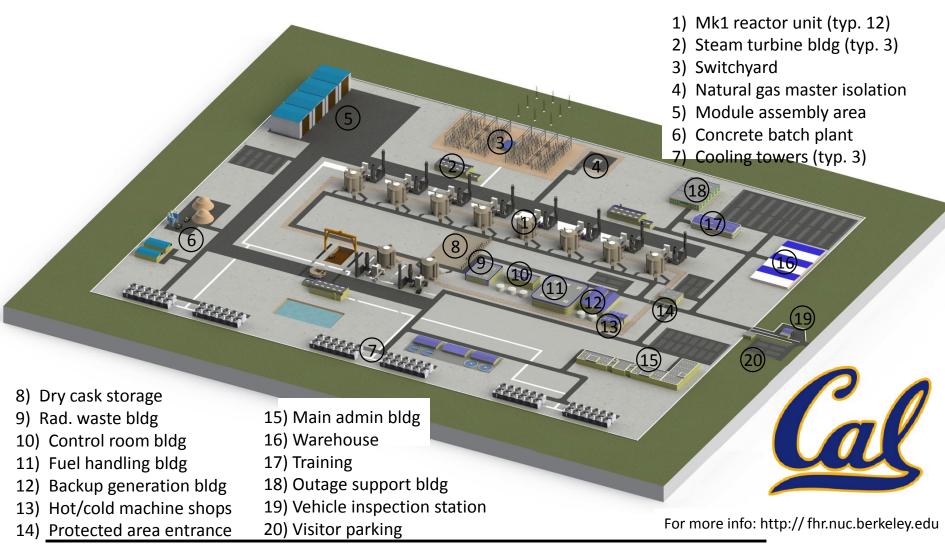
### **List of IRP-2 Tasks**

- FHR Tritium Control and the Role of Carbon (Largest Task)
- FHR Corrosion Control with Redox Control, Impurity Control, and Materials Selection
- FHR Experiments and Modeling for Thermal Hydraulics, Neutronics and Structural Mechanics
- FHR Evaluation Model Benchmarking and Validation Workshops
- Using Lessons Learned From FHR R&D to Advance All Generation IV Technologies



#### Notional 12-unit Mk1 PB-FHR nuclear station

1200 MWe base load; 2900 MWe peak





## FHRs differ from other reactor classes in several key ways

- FHR fuel reaches full depletion in a short period of time
- Primary system is compact compared to HTGRs and SFRs
- Core fissile inventory is remarkably small
- Core Cs-137 inventory is remarkably small
- Uranium and enrichment requirements similar to LWRs and HTGRs

		Westing-				
		ORNL	house			
	Mk1	2012	4-loop		S-	
	PB-FHR	AHTR	PWŔ	<b>PBMR</b>	PRISM	
MWt)	236	3400	3411	400	1000	

PWR HTGR SFR

**FHRs** 

		Mk1 PB-FHR	ORNL 2012 AHTR	house 4-loop PWR	PBMR	S- PRISM
	Reactor thermal power (MWt)	236	3400	3411	400	1000
	Reactor electrical power (MWe)	100	1530	1092	175	380
	Fuel enrichment †	19.90%	9.00%	4.50%	9.60%	<b>8.93%</b>
	Fuel discharge burn up (MWt-d/kg)	180	71	48	92	106
7	Fuel full-power residence time in core (yr)	1.38	1.00	3.15	2.50	7.59
	Power conversion efficiency	42.4%	45.0%	32.0%	43.8%	38.0%
	Core power density (MWt/m3)	22.7	12.9	105.2	4.8	321.1
	Fuel average surface heat flux (MWt/m2)	0.189	0.285	0.637	0.080	1.13
	Reactor vessel diameter (m)	3.5	10.5	6.0	6.2	9.0
L	Reactor vessel height (m)	12.0	19.1	13.6	24.0	20.0
	Reactor vessel specific power (MWe/m3)	0.866	0.925	2.839	0.242	0.299
Ł	Start-up fissile inventory (kg-U235/MWe) ††	0.79	0.62	2.02	1.30	6.15
	EOC Cs-137 inventory in core (g/MWe) *	30.8	26.1	104.8	53.8	269.5
	EOC Cs-137 inventory in core (Ci/MWe) *	2672	2260	9083	4667	23359
	Spent fuel dry storage density (MWe-d/m3)	4855	2120	15413	1922	-
	Natural uranium (MWe-d/kg-NU) **	1.56	1.47	1.46	1.73	-
	Separative work (MWe-d/kg-SWU) **	1.98	2.08	2.43	2.42	-
	LE CERTICAL CO. C. C. L.	66 1 114		1 D 1 C 1		

<sup>†</sup> For S-PRISM, effective enrichment is the Beginning of Cycle weight fraction of fissile Pu in fuel



<sup>††</sup> Assume start-up U-235 enrichment is 60% of equilibrium enrichment; for S-PRISM startup uses fissile Pu

<sup>\*</sup> End of Cycle (EOC) life value (fixed fuel) or equilibrium value (pebble fuel)

<sup>\*\*</sup> Assumes a uranium tails assay of 0.003.

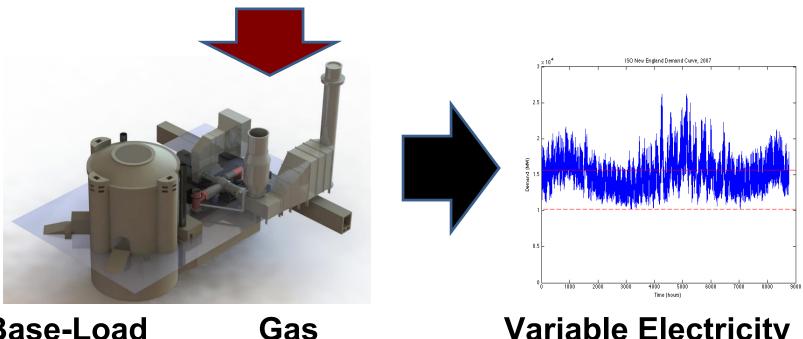
# FHRs have unique safety characteristics for accidents resulting in long-term off-site land use restrictions from Cs-137

	FHRs	LWRs		
Low Cs-137 inventory	~30 g/MWe	~105 g/MWe		
High thermal margin to fuel damage	T <sub>damage</sub> > 1800°C	T <sub>damage</sub> ~ 830 – 1250°C		
High solubility of cesium in coolant	CsF has high solubility	Cs forms volatile compounds		
Intrinsic low pressure	High coolant boiling temperature and chemical stability	High vapor pressure at accident temperatures		



## FHR Couples with Nuclear Air-Brayton Combined Cycle (NACC) and Firebrick Resistance-Heated Energy Storage (FIRES)

**Stored Heat and/or Natural Gas** 



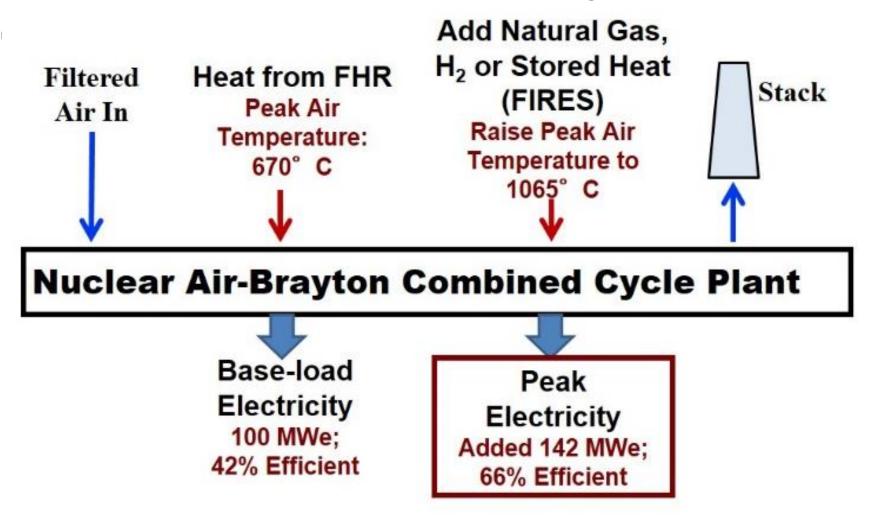
Base-Load Reactor

**Gas Turbine**  Variable Electricity
And Steam

Base-load Reactor with Power Station that Buys or Sells Electricity As Needed

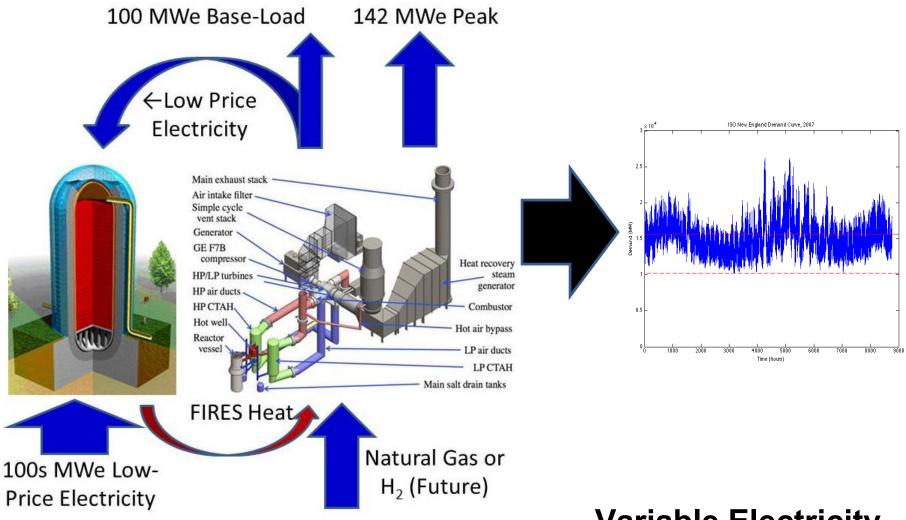
## **Gas Turbine Operates in Two Modes**

**NACC** for Variable Electricity Output



Topping Cycle: 66% Efficient for added Heat-to-Electricity: Stand-Alone Natural Gas Plants 60% Efficient

# FHR With NACC Can Incorporate Firebrick Resistance-Heated Energy Storage (FIRES)



**Base-Load Reactor, NACC and FIRES** 

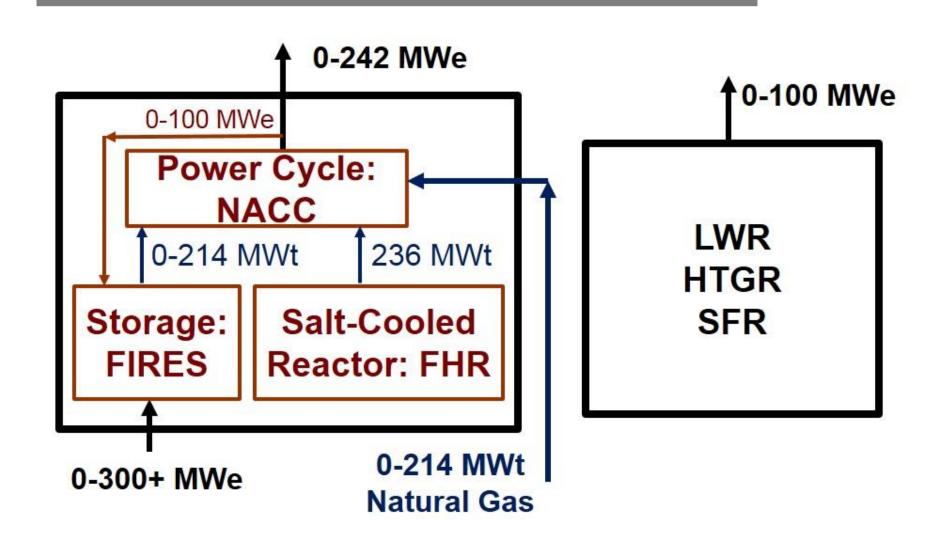
Variable Electricity
And Steam 3

#### **Economic Basis for FHR with NACC**

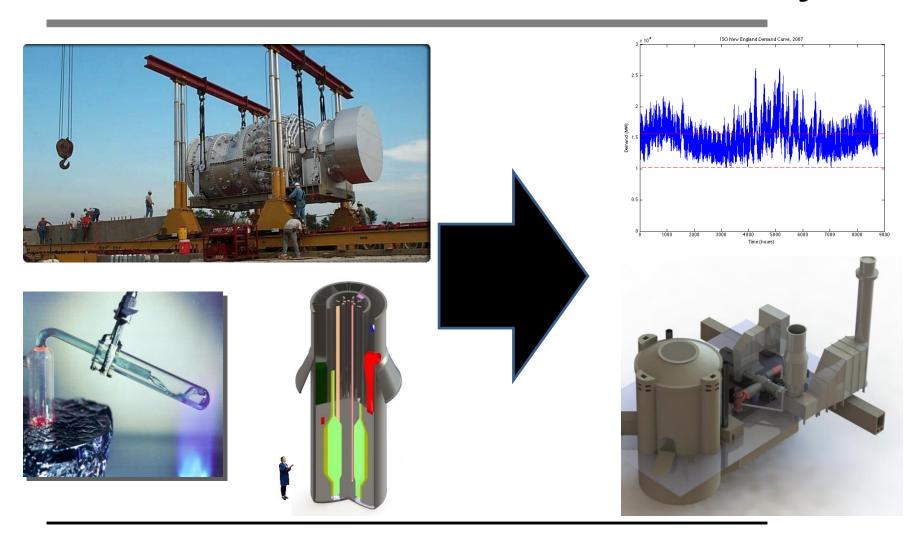
- Competing with Natural Gas (NG)
  - Base-load heat-to-electricity efficiency: 42%
  - Peak electricity with incremental NG gas efficiency: 66%
  - For peak electricity, more efficient than stand-alone NG plants (60%) and thus 50% increase in revenue over baseload-only nuclear plants after pay for NG
- Competing with renewables and enabling a low-carbon nuclear renewable grid
  - At times of low prices (excess electricity) convert electricity to high-temperature stored heat using Firebrick resistance-Heated Energy Storage (FIRES)
  - FIRES heat replaces burning of natural gas
  - Converting low-price electricity to high-price electricity



### From the Grid Perspective, FHR/NACC/FIRES is a Second Class of Nuclear Power



# FHR With Nuclear Air Brayton Cycle and FIRES Creates a Second Class of Nuclear Power Systems



Separate from LWR/SFR/HTGR That Compete for Same Energy Market

# **Beryllium Safety for Flibe Work**

- 1. Operated a flibe laboratory since 2012: walk-in fume hood, and glove-boxes. Purification, salt-loop, salt transfers, and glove-box experiments.
- 2. Additional hazards associated with flibe handling: HF, F2, high temperature, voltage, challenges of salt transfer while ensuring purity.
- 3. Ensuring inert atmosphere goes hand in hand with containing Be

FLIBE RESEARCHERS AT UW. JAN 2016, NEXT TO FLIBE GLOVE-BOX

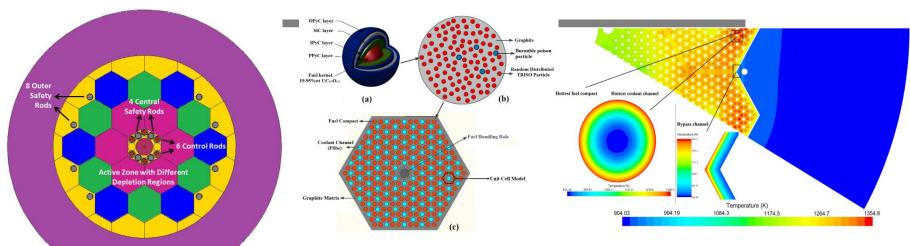


FLIBE WALK-IN FUME-HOOD WITH HF PURIFICATION



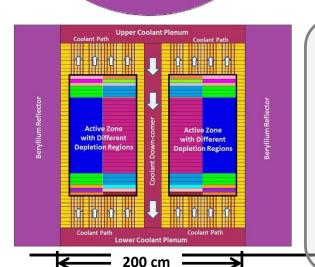
- 4. Air monitoring: Met OSHA PEL: 2 ug/m3 and Action Level: 0.2 ug/m3
- 5. Surface swipes housekeeping: 3 ug/100 cm2, general release: 0.2 ug/100 cm2. Occasional swipes above 0.2 ug/100 cm2 were followed by clean-up to ensure good housekeeping.
- 6. Better understanding of source term and particulate size distribution for flibe activity would be valuable
- Additional options for real-time beryllium monitoring, and health monitoring should continue to be explored

## 10 MWth Transportable FHR (TFHR)



**Prismatic assembly with TRISO fuel particles** 

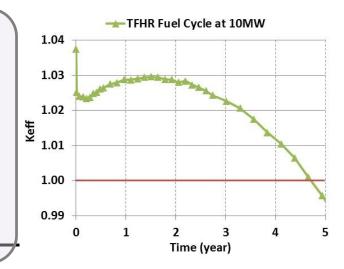
**Full 3-dimensional CFD modeling** 



**Beryllium Reflector** 

#### **Design Features**

- > 10 MWth with ~ 5-yr fuel cycle
- Compact core ~ 2-m diameter
- Tranportable by air, rail or truck
- Flibe salt coolant 600-700 ° C
- > High efficiency air Brayton cycle
- 18 prismatic fuel assemblies
- 6 control rods and 12 safety rods
- Center coolant down-comer
  - > 10-year fuel cycle optimization in progress



**Massachusetts Institute of Technology**